

This article was downloaded by:

On: 24 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Journal of Liquid Chromatography & Related Technologies

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597273>

### Application of Thin-Layer Chromatography to High-Performance Liquid Chromatographic Separation of Steroidal Hormones and Cephalosporin Antibiotics

T. Okumura<sup>a</sup>

<sup>a</sup> Shionogi Research Laboratories, Shionogi & Co., Ltd., Osaka, Japan

**To cite this Article** Okumura, T.(1981) 'Application of Thin-Layer Chromatography to High-Performance Liquid Chromatographic Separation of Steroidal Hormones and Cephalosporin Antibiotics', *Journal of Liquid Chromatography & Related Technologies*, 4: 6, 1035 – 1064

**To link to this Article:** DOI: 10.1080/01483918108059603

**URL:** <http://dx.doi.org/10.1080/01483918108059603>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

APPLICATION OF THIN-LAYER CHROMATOGRAPHY TO  
HIGH-PERFORMANCE LIQUID CHROMATOGRAPHIC SEPARATION  
OF STEROIDAL HORMONES AND CEPHALOSPORIN ANTIBIOTICS

T. Okumura

Shionogi Research Laboratories, Shionogi & Co., Ltd.,  
Fukushima-ku, Osaka 553, Japan

ABSTRACT

High-performance liquid chromatographic (HPLC) separation of steroidal hormones and cephalosporin antibiotics was investigated by adsorption chromatography and reversed-phase chromatography, respectively.

Prior to the HPLC separation of these pharmaceuticals, silica gel thin-layer adsorption chromatography of steroidal hormones and reversed-phase thin-layer partition chromatography of cephalosporin antibiotics with chemically bonded dimethylsilyl silica gel were performed in order to obtain suitable HPLC separation systems.

In the separation of steroidal hormones, the same binary mobile phase ratios of TLC did not give satisfactory results in HPLC. For the sharp separation in HPLC, solvent strength in the binary solvent mixture used for TLC had to be decreased.

The difference in solvent strength for efficient separation between TLC and HPLC might be

attributed to the fact that in HPLC the solvent elution power acts in an isocratic manner while in TLC it acts in a gradient manner.

On the other hand, a correlation of mobility between TLC and HPLC separation for cephalosporin antibiotics was obtained, and the possibility of direct transfer of chromatographic systems from TLC to HPLC for separation of these antibiotics was confirmed.

### INTRODUCTION

Several authors have reported the relationship between TLC and classical column chromatography of various kinds of compounds (1-4). Recently, similar studies between TLC and HPLC separations have been described (5-9).

This study used a TLC separation as a pilot study for HPLC analysis for steroidal hormones (adsorption chromatography) and cephalosporin antibiotics (reversed-phase partition chromatography). The HPLC separation of six hardly resolvable pairs of steroidal steric and geometrical isomers shows the parallelism with the TLC separation and the requirement of a lower solvent strength of the binary solvent mixture used for TLC. The analysis of eight kinds of cephalosporin antibiotics confirmed a correlation between separation by TLC and HPLC and the

possibility of direct transfer of chromatographic systems from TLC to HPLC.

#### MATERIALS

Commercially available Merck silica gel 60 F<sub>254</sub> (Item No. 5715) and silica gel 60 F<sub>254</sub> silanised (Item No. 5747, laboratory-prepared) were used after activation at 110°C for 30 min. Waters Model 204 Liquid chromatograph with Models 6000 A Solvent Delivery System and U6K Universal Injector was used for HPLC analysis. An isocratic mode was employed and the eluent was monitored at 240 nm for steroidal hormones and 265 nm for cephalosporin antibiotics with Waters Model 440 ultraviolet absorbance detector or Shimadzu SPD-2A variable wavelength UV monitor in series with Waters Model R-401 Difference Refractometer. Ultraviolet and refractometry index chromatograms were recorded with Shimadzu Model C-R1A Chromatopac computing recorder and integrator. Chromatographic conditions are described in the figure legends. Merck LiChrosorb RP-2 [10 µm, 4 mm x 30 cm, laboratory-packed by balanced density slurry method (10-11)], LiChrosorb SI 100 (10 µm, 4 mm x 30 cm) and Waters µPorasil (7 µm, 4 mm x 30

cm) were used for analysis. The solvents used were filtered with Milipore membrane filter (0.45  $\mu\text{m}$ ) and degassed. As an ion-pair reagent, Waters PIC B-7 [1-heptane sulfonic acid (pH 3.5 with acetic acid)] and PIC A (tetrabutylammonium phosphate, pH 7.5) were used for the cephalosporin analysis.

## RESULTS AND DISCUSSION

### Chromatography of Steroidal Hormones (11)

As pairs of stereoisomers, 5:6-oxido-cholester-yl benzoates ( $\alpha = 1$ ,  $\beta = 2$ , 12), 5:10-oxido-19-norandrostane-3 $\alpha$ ,17 $\beta$ -diols ( $\alpha = 3$ ,  $\beta = 4$ , 12) and its 3,5-dinitrobenzoates (DNB), dexamethasone 21-acetate (16 $\alpha$ -methyl = 5), betamethasone 21-acetate (16 $\beta$ -methyl = 6), 5-cyano-19-nortestosterone 3,5-DNB ( $\alpha$ -cyano = 9,  $\beta$ -cyano = 10, 13) were used. As a pair of geometrical isomers, 3 $\beta$ -hydroxy-androsten-17-one oxime benzoate (syn = 7, anti = 8, 14) was used.

TLC separation of these steroidal isomers was performed with solvent systems having the same solvent strength (15) such as (A) benzene-*i*-propyl ether (1:1), (B) chloroform-ethyl acetate (9:1), (C) benzene-methanol (19:1) and (D) chloroform-acetone

TABLE 1  
TLC of Steroidal Cyanoketone Stereoisomers<sup>a)</sup>

Mobile phase	$\alpha$ -Cyanoketone	$\alpha$	$\beta$ -Cyanoketone	Results
Benzene-isopropyl ether (1:1)	51	1.21	42	Separable
Chloroform-ethyl ether (9:1)	46	1.15	40	Separable
Benzene-methanol (19:1)	38	1.00	38	Inseparable
Chloroform-acetone (20:1)	32	1.00	32	Inseparable

a) Stereoisomer: 5-Cyano-19-nortestosterone 3,5-dinitrobenzoate

TLC plate: Merck silica gel 60 F<sub>254</sub>

Detection: Fluorescence quenching at 254 nm followed by sulfuric acid

$\alpha$ : Separation factor = larger  $hR_f$ /smaller  $hR_f$

(20:1). For example, 5-cyano-19-nortestosterone isomers were separable with benzene-*i*-propyl ether (1:1) or chloroform-ethyl acetate (9:1) (Table 1 and Figure 1). Table 3 shows the favorable TLC condition for the six kinds of steroidal isomers.

HPLC separation of these steroidal isomers was carried out under the same conditions used for TLC.

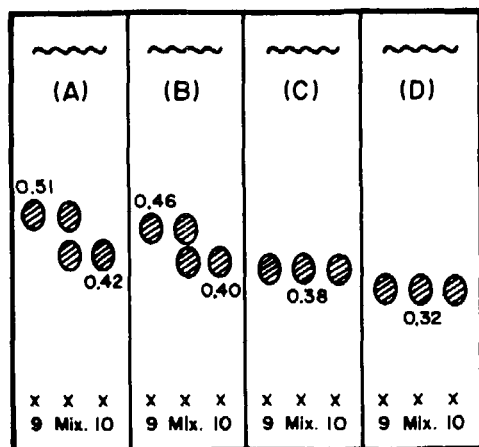


Figure 1. TLC of Steroidal Cyanoketone Stereoisomers (9) and (10)  
 Solvent : (A) benzene-*i*-propyl ether (1:1),  
 (B) chloroform-acetone (9:1),  
 (C) benzene-methanol (19:1), (D) chloroform-acetone (20:1); Plate : Merck silica gel 60 F<sub>254</sub>; Detection : fluorescence quenching at 254 nm followed by H<sub>2</sub>SO<sub>4</sub>

### Relationship between Mobile Phase and Retention Times

A preliminary HPLC separation of 5:6-oxido-cholesteryl 3-benzoates (1 and 2) was tried (Table 2). The chromatographic behaviors showed that the favorable TLC mobile phase ratios gave insufficient HPLC separation for the steroidal isomers. In order to obtain good HPLC separation of these stereoisomers, the  $hR_f$  values had to be lowered by controlling the solvent strength in the mobile phases. Table 3 and Figure 2 show the results of favorable HPLC separation for the cholesteryl epoxides.

### HPLC Separations of Other Steroidal Isomers

Figures 3, 4, 5 and 6 show the HPLC separation of steroidal stereoisomers such as 3 and 4, 5 and 6, 9 and 10, and geometrical isomers (7 and 8). Table 4 summarize favorable TLC and HPLC conditions for these steroidal compounds.

### Chromatography of Cephalosporin Antibiotics (15)

To separate the cephalosporin antibiotics, 7-aminodesacetoxycephalosporanic acid (7-ADCA),



TABLE 2  
HPLC of Cholesteryl Oxide Isomer<sup>a)</sup>

Mobile phase <sup>b)</sup>	Flow rate (ml/min)	Pressure (psi)	t <sub>R</sub> α-Oxide	t <sub>R</sub> β-Oxide	t <sub>O</sub>	α	Result
n-C <sub>6</sub> H <sub>14</sub> -CHCl <sub>3</sub> (1:4)	2.0	1100	1.5	1.2	1.5	1.00	Inseparable
	1.5	750	2.0	1.7	2.0	1.00	"
	1.0	550	3.2	2.7	3.2	1.00	"
	0.5	250	6.5	5.6	6.5	1.00	"
n-C <sub>6</sub> H <sub>14</sub> -CHCl <sub>3</sub> (1:1)	2.0	1000	2.0	1.6	1.8	2.00	separable
	1.5	700	2.6	2.1	2.3	2.51	"
	1.0	480	3.7	3.1	3.6	1.20	"
	0.5	200	7.5	6.3	6.9	2.00	"

a) Stereoisomer: 5:6-Oxido-cholesteryl 3-benzoate

HPLC instruments: Waters 6000 A pump and U6K injector, Shimadzu SPD-2A variable wavelength UV monitor at 240 nm and 0.2 AUFS, Merck LiChrosorb SI 100 (10 μm, 4 mm x 30 cm), Shimadzu Chromatopak C-RIA computing recorder and integrator.

b) Not eluted with n-C<sub>6</sub>H<sub>14</sub>

TABLE 3  
Favorable TLC hr<sub>f</sub> Value for HPLC Separation

Compound	Mobile phase	TLC hr <sub>f</sub> <sup>a)</sup>		TLC hr <sub>f</sub> <sup>b)</sup>		HPLC hr <sub>f</sub> <sup>c)</sup>	
		α	β	α	β	α	β
5:6-Oxido-cholesteryl 3-benzoate	CHCl <sub>3</sub> -n-C <sub>6</sub> H <sub>14</sub> (1:1)	11	21	15	25	84	91
5:10-Oxido-19-nor-androstane-3α,17β-diol	CHCl <sub>3</sub> -CH <sub>3</sub> OH (20:1)	6	5	26	21	51	42
Dexamethasone 21-Ac Betamethasone 21-Ac	"	11	6	11	8	103	95
3β-Hydroxy-5-androsten-17-one 3-benzoate	Benzene-i-PrOH (4:1)	syn 22	anti 11	syn 27	anti 18	syn 55	anti 43
5-Cyano-19-nor-testosterone 3,5-dinitrobenzoate	"	19	16	28	24	39	34
Cephalosporin d) stereoisomer	CHCl <sub>3</sub> -CH <sub>3</sub> OH (50:1)	A 6	B 6	A 18	B 13	A 118	B 94

Merck TLC plate silica gel 60 F<sub>254</sub>, a) solvent oversaturation, b) solvent normal saturation.

c) Merck LiChrosorb SI 100 (10 μm, 4 mm x 30 cm), hr = 100 x t<sub>o</sub>/t<sub>R</sub>.

d) The chemical structures were not identified.

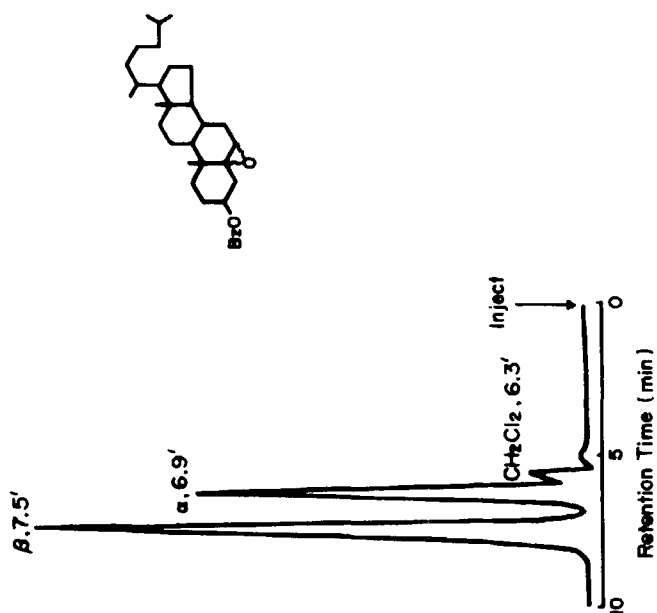


Figure 2. LC of cholesterol epoxides  
 Column,  $\mu\text{Porasil}$  ( $10\ \mu\text{m}$ ,  $4\ \text{mm} \times 30\ \text{cm}$ );  
 carrier,  $\text{CHCl}_3$ -*n*-hexane (1:1), isocratic;  
 flow rate,  $2\ \text{ml}/\text{min}$ ; pressure,  $1100\ \text{psi}$ ;  
 detector, UV220  $0.2\ \text{AUFS}$ ;  
 sample size,  $10\ \mu\text{g}/\mu\text{l}$  in  $\text{CH}_2\text{Cl}_2$   
 Temperature, ambient

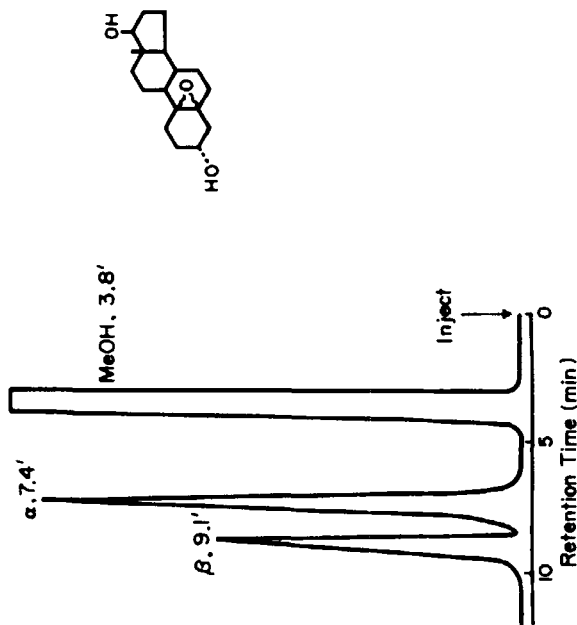


Figure 3. LC of steroidal 5,10-epoxides  
 Column,  $\mu\text{Porasil}$  ( $10\ \mu\text{m}$ ,  $4\ \text{mm} \times 30\ \text{cm}$ );  
 carrier,  $\text{CHCl}_3$ - $\text{MeOH}$  (100 : 5);  
 flow rate,  $1\ \text{ml}/\text{min}$ ; pressure,  $500\ \text{psi}$ ;  
 detector, RI 8X;  
 sample size,  $100\ \mu\text{g}/10\ \mu\text{l}$  in  $\text{MeOH}$

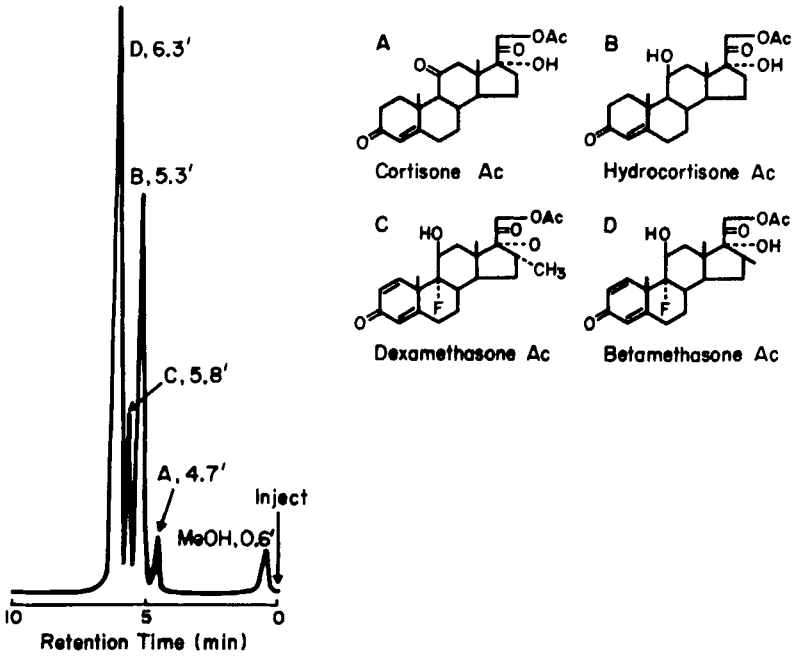


Figure 4. LC of steroidal hormones  
 Column,  $\mu$ Porasil ( $10\ \mu\text{m}$ ,  $4\text{mm} \times 30\text{cm}$ );  
 carrier,  $\text{CHCl}_3\text{-MeOH}$  (100:5);  
 flow rate;  $0.6\ \text{ml/min}$ ; pressure,  $500\ \text{psi}$ ;  
 detector, UV240  $0.2\ \text{AUFS}$ ;  
 sample size, A.  $2\ \mu\text{g}$  B.  $8\ \mu\text{g}$  C.  $5\ \mu\text{g}$   
 D.  $10\ \mu\text{g}/\mu\text{l}$  in MeOH

cephalexin (CEX), 7-aminocephalosporanic acid (7-ACA), cephalothin (CET), cephaloglycine (CEG), desacetylcephalothin (DACET), ethoxycarbonyl DACET and cephaloridine (CER) were used.

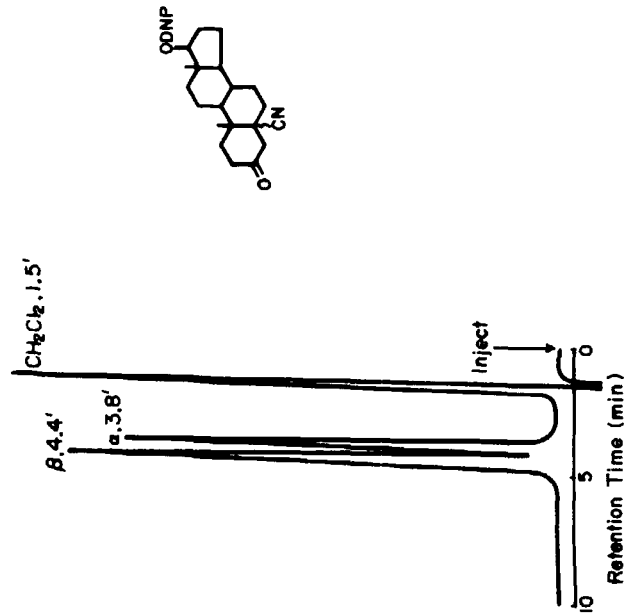


Figure 5. LC of steroidal 17-ketoximes  
 Column,  $\mu$  Porasil (10  $\mu$ m, 4mm x 30cm);  
 carrier, benzene-IPE (4:1);  
 flow rate, 1ml/min; pressure, 600 psi;  
 detector, RI 8X;  
 sample size, 100  $\mu$ g/5  $\mu$ l in  $\text{CH}_2\text{Cl}_2$

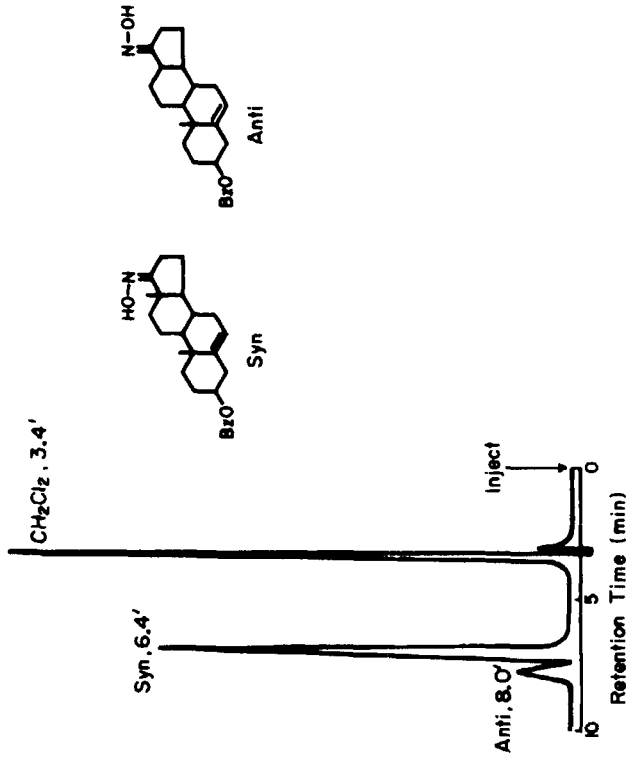


Figure 6. LC of steroidal cyanoketone  
 Column,  $\mu$  Porasil (10  $\mu$ m, 4mm x 30cm);  
 carrier, benzene-IPE (4:1);  
 flow rate, 2ml/min; pressure, 1200 psi;  
 detector, RI 8X;  
 sample size, 100  $\mu$ g/10  $\mu$ l in  $\text{CH}_2\text{Cl}_2$

TABLE 4  
Favorable TLC and HPLC Condition for Steroid Compounds<sup>a)</sup>

Steroid	Mobile phase	Solvent ratio HPLC	Solvent ratio TLC
5:6-Oxido-cholesteryl 3-benzoate	CHCl <sub>3</sub> -n-C <sub>6</sub> H <sub>14</sub>	(1:1)	(4:1)
5:6-Oxido-19-nor-androstane- 3 $\alpha$ , 17 $\beta$ -diol	CHCl <sub>3</sub> -CH <sub>3</sub> OH	(20:1)	(10:1)
Dexamethasone 21-acetate Betamethasone 21-acetate	CHCl <sub>3</sub> -CH <sub>3</sub> OH	(20:1)	(10:1)
3 $\beta$ -Hydroxy-5-androsten- 17-one oxime 3-benzoate	Benzene-isopropyl ether	(4:1)	(2:1)
5-Cyano-19-nor-testosterone 3,5-dinitrobenzoate	Benzene-isopropyl ether	(4:1)	(2:1)
Cephalosporin stereoisomer	CHCl <sub>3</sub> -CH <sub>3</sub> OH	(50:1)	(20:1)

a) TLC and HPLC condition as in TABLES 1-3.

Reversed-Phase TLC (RPTLC) of Cephalosporin Anti-biotics

The author devised an eluotropic series for reversed-phase chromatography prior to RPTLC (Table 5). From this series, methanol-water, acetone-water, dioxane-water, acetonitrile-water and tetrahydrofuran-water were selected for RPTLC of the cephalosporins, 7-aminocephalosporanic acid, cephalothin,

TABLE 5

Eluotropic Series for Reversed-Phase Chromatography

Solvent	Polarity index
Water	9.0
Methanol	6.6
Acetic acid	6.2
Ethanol	5.2
Acetonitrile	6.2
2-Propanol	4.3
N,N-Dimethylformamide	6.4
Acetone	5.4
<u>n</u> -Propanol	3.9
Dioxane	4.8

cephalexin, cephaloglycine and cephaloridine were carried out (Table 6). Comparison of the mean separation factor  $\bar{\alpha}$  (larger  $hR_f$  value/smaller  $hR_f$  value) gave methanol-water (1:4) as the optimum condition. Among the mixing ratios of methanol-water of 1:1, 1:2 and 1:4, the last was the most effective (Table 7). Eight kinds of cephalosporin were separable using this mobile phase (Figure 7).

#### Ion-Pair RPTLC of Cephalosporin Antibiotics

In order to decrease the  $hR_f$  values of 7-aminocephalosporanic acid, cephalothin and cephaloglycine, ion-pair chromatography was applied to RPTLC of the cephalosporins (Tables 8 and 9). The ion-pair chromatographic (16) effect was recognized in the separation of cephaloglycine, cephalexin and cephaloridine [mobile phase; methanol-water (1:1) with PIC B-7], 7-aminocephalosporanic acid and cephalothin [methanol-water (1:4) with PIC B-7] (Table 10), and also in the separation of 7-aminocephalosporanic acid, cephaloglycine, cephalexin and cephalothin [methanol-water (1:2) with PIC A] (Table 11). Thus, ion-pair chromatography was found to be useful for cephalosporin compounds and amino acid.



TABLE 6  
Solvent Effect in Reversed-Phase TLC of Cephalosporin Antibiotics a)

Compound	MeOH-H <sub>2</sub> O (1:4)		Dioxane-H <sub>2</sub> O (1:4)		AN-H <sub>2</sub> O (1:4)		CH <sub>3</sub> CN-H <sub>2</sub> O (1:4)		THF-H <sub>2</sub> O (1:4)	
	hR <sub>f</sub>	$\alpha$	hR <sub>f</sub>	$\alpha$	hR <sub>f</sub>	$\alpha$	hR <sub>f</sub>	$\alpha$	hR <sub>f</sub>	$\alpha$
7-Aminocephalo- sporanic acid	93		96		94		89		92	
	1.33		1.37		1.38		1.31		1.37	
Cephalothin	70		49		54		53		49	
	1.95		1.21		1.26		1.19		1.08	
Cephalexin	61		70		68		68		67	
	1.17		1.18		1.23		1.08		1.27	
Cephaloglycine	52		58		42		57		62	
	1.49		1.23		1.05		1.26		1.11	
Cephaloridine	35		40		44		42		44	
Mean separation factor $\alpha$	1.28	>	1.25	>	1.25	>	1.21	>	1.21	>
Developing rate (min/10 cm)	55		50		50		45		75	

a) TLC plate: Merck silica gel 60 HF<sub>254</sub> silanised

Detection: Fluorescence quenching followed by I<sub>2</sub> vapor

hR<sub>f</sub> = 100 x R<sub>f</sub>,  $\alpha$  = larger hR<sub>f</sub>/smaller hR<sub>f</sub>

AN: acetone, THF: tetrahydrofuran

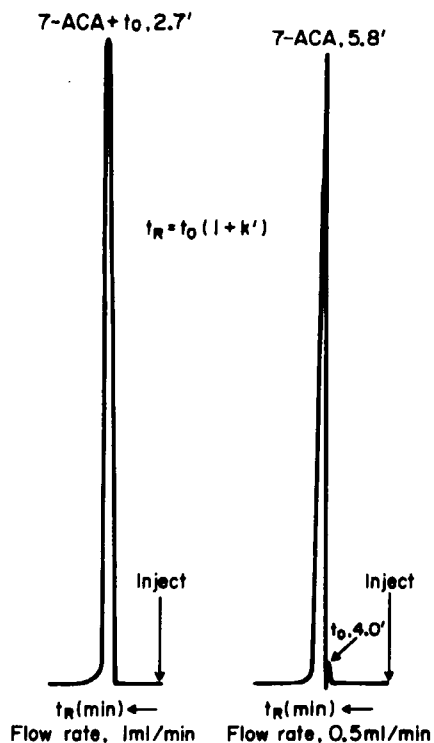


Figure 7. Simultaneous Reversed Phase TLC of Cephalosporins  
Solvent, MeOH-H<sub>2</sub>O (1:4); Plate, Silica gel 60 HF<sub>254</sub>  
silanised; Detection, UV<sub>254</sub>-quenching followed by I<sub>2</sub> vapor.

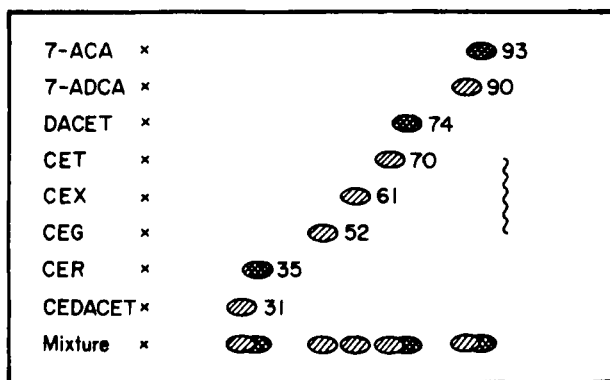


Figure 8. Influence of Flow rate on Separation of Cephalosporin  
Column, RP-2 (4x30cm); Mobile Phase, MeOH-H<sub>2</sub>O (1:4)

TABLE 7

Reversed-Phase TLC of Cephalosporin Antibiotics<sup>a)</sup>

Compound	hR <sub>f</sub> values of cephalosporins			
	MW11	MW12	MW14	H <sub>2</sub> O <sup>b)</sup>
7-Aminocephalo- sporanic acid	93	92	93	55
Cephaloglycine	81	64	52	15
Cephalexin	81	71	61	34
Cephaloridine	68	51	35	6
Cephalothin	83	76	70	31
Developing rate (min/10 cm)	40	35	53	65/7 cm

a) TLC plate: Merck silica gel HF<sub>254</sub> silanised  
(laboratory-prepared)

Detection: Fluorescence quenching, I<sub>2</sub> vapor,  
Fluorescamine and Ninhydrin reagents

MW = MeOH-H<sub>2</sub>O

b) Abnormal development due to the lipophilic  
adsorbent

Reversed-Phase HPLC (RPHPLC) of Cephalosporin  
Antibiotics

Four kinds of mobile phases used in RPTLC  
separation, methanol-water, tetrahydrofuran-water,  
acetonitrile-water and dioxane-water, were applied

TABLE 8  
 Ion-Pair Reversed-Phase TLC of Cephalosporin  
 Antibiotics<sup>a)</sup>

Compound	hR <sub>f</sub> values of cephalosporins <sup>b)</sup>			
	MW11B	MW12B	MW14B	H <sub>2</sub> O <sub>B</sub>
7-Aminocephalo- sporanic acid	89	85	81	42
Cephaloglycine	71	56	49	17
Cephalexin	70	65	55	23
Cephaloridine	58	48	37	15
Cephalothin	75	57	46	18
Developing rate (min/10 cm)	35	35	35	90

a) To each mobile phase, 0.005 M 1-heptanesulfonic acid-ACOH (pH 3.5, B) was added.

b) TLC plate and detection as in TABLE 7.

to RPHPLC of cephalosporin antibiotics (Table 12). Among these mobile phases, methanol-water (1:4) was the most effective, as for RPTLC separation (Figure 9). The 1:1 and 1:2 ratios of methanol-water gave three peaks [7-aminocephalosporanic acid and cephalothin, cephalexin and cephaloglycine, and cephaloridine]. The 1:4 ratio gave four peaks [7-

TABLE 9  
Ion-Pair Reversed-Phase TLC of Cephalosporin  
Antibiotics<sup>a)</sup>

Compound	hR <sub>f</sub> values of cephalosporins <sup>b)</sup>			
	MW11A	MW12A	MW14A	H <sub>2</sub> OA
7-Aminocephalo- sporanic acid	81	80	77	75
Cephaloglycine	78	50	43	17
Cephalexin	79	61	64	29
Cephaloridine	60	46	32	10
Cephalothin	78	59	49	22
Developing rate (min/10 cm)	40	50	40	60

a) To each mobile phase, 0.005 M tetrabutylammonium phosphate (pH 7.5, A) was added.

b) TLC plate and detection as in TABLE 7.

TABLE 10  
Ion-Pair TLC Effect on Cephalosporin Antibiotics

Compound	$\Delta$ hR <sub>f</sub> values of cephalosporins <sup>a)</sup>			
	MW11B	MW12B	MW14B	H <sub>2</sub> OB
7-Aminocephalo- sporanic acid	4	7	12	13
Cephaloglycine	10	8	3	-2
Cephalexin	11	6	6	11
Cephaloridine	10	3	-2	-9
Cephalothin	8	21	24	13

a)  $\Delta$ hR<sub>f</sub> = hR<sub>f</sub> in TABLE 8 - hR<sub>f</sub> in TABLE 7.

TABLE 11

## Ion-Pair TLC Effect on Cephalosporin Antibiotics

Compound	$\Delta hR_f$ values of cephalosporins <sup>a)</sup>			
	MW11A	MW12A	MW14A	H <sub>2</sub> OA
7-Aminocephalosporanic acid	12	12	16	-20
Cephaloglycine	3	14	8	-2
Cephalexin	2	10	-3	5
Cephaloridine	8	5	3	-4
Cephalothin	5	17	21	9

a)  $\Delta hR_f = hR_f$  in TABLE 9 -  $hR_f$  in TABLE 7.

aminocephalosporanic acid and cephalothin, cephalixin, cephaloglycine, and cephaloridine (Table 9).

#### Influence of Flow Rate

Three flow rates, 1 ml/min, 0.5 ml/min and 0.3 ml/min were used for the HPLC separation [stationary phase, Merck RP-2 home-packed column; mobile phase, methanol-water (1:4)] (Figure 8). A flow rate of 0.5 ml/min resulted in separation of dead volume ( $t_0$ ) and 7-aminocephalosporanic acid.

TABLE 12  
Solvent Effect in Reversed-Phase HPLC of Cephalosporin Antibiotics<sup>a)</sup>

Compound	MeOH-H <sub>2</sub> O(1:4) t <sub>R</sub> k' α	THF-H <sub>2</sub> O(1:4) t <sub>R</sub> k' α	CH <sub>3</sub> CN-H <sub>2</sub> O(1:4) t <sub>R</sub> k' α	Dioxane-H <sub>2</sub> O(1:4) t <sub>R</sub> k' α
7-Aminocephalo- sporanic acid	5.8 0.07 1.45	5.2 0.04 3.00	4.6 -0.04	5.6 0.14
Cephalothin	6.0 0.10 6.67	5.6 0.12 2.17	4.8 0.00	6.0 0.23 1.57
Cephaloglycine	9.0 0.67 1.30	6.3 0.26 1.69	6.0 0.25 1.17	6.8 0.39 1.72
Cephalexin	8.2 0.52 3.07	7.2 0.44 1.55	6.2 0.29 3.35	7.2 0.47 1.21
Cephaloridine	14.0 1.60	8.4 0.63	9.5 0.98	9.0 0.84 1.79
t <sub>0</sub>	5.4	5.0	4.8	4.9
Column pressure (psi)	1600	2500	1600	2100
Separation	4 peaks >	4 peaks >	3 peaks >	3 peaks >

a) HPLC instruments: Waters ALC/GPC 204, Shimadzu SPD-2A variable wavelength UV monitor at 265 nm and 0.1 AUFS  
Column: Merck RP-2 home-packed by balanced density slurry method (4 mm x 30 cm)

Chart speed: 0.5 cm/min

Sample size: 0.25 µg/µl in NaHCO<sub>3</sub>-HCl buffer (pH 6.6)

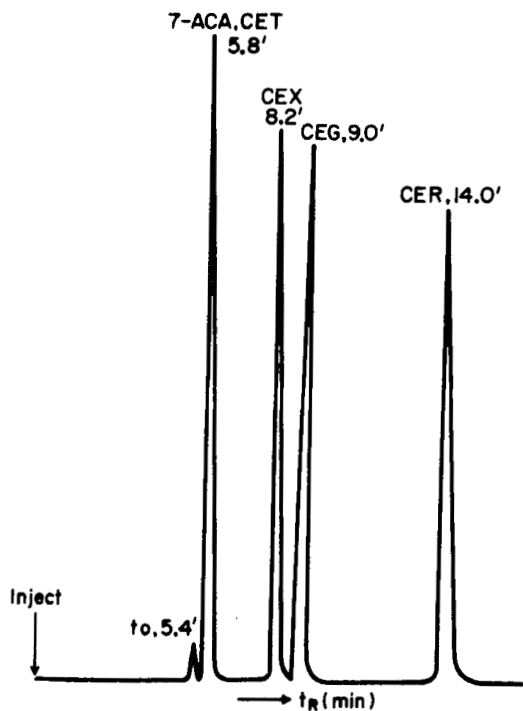


Figure 9. LC of Cephalosporins

Column, RP-2 (4×30 cm); Mobile Phase, MeOH-H<sub>2</sub>O (1:4);  
Flow Rate, 0.5 ml/min; Column Pressure, 1600 psi.

#### Ion-Pair RPHPLC of Cephalosporin Antibiotics

7-Aminocephalosporanic acid and cephalothin could not be separated without addition of PIC reagent (Table 12). By using methanol-water (1:2) with PIC B-7 and A as for the RPTLC separation, five cephalosporins, 7-aminocephalosporanic acid, cephaloglycine, cephaloridine, cephalothin and cephalixin, could be separated from each other (Figures 10, 11 and 12).



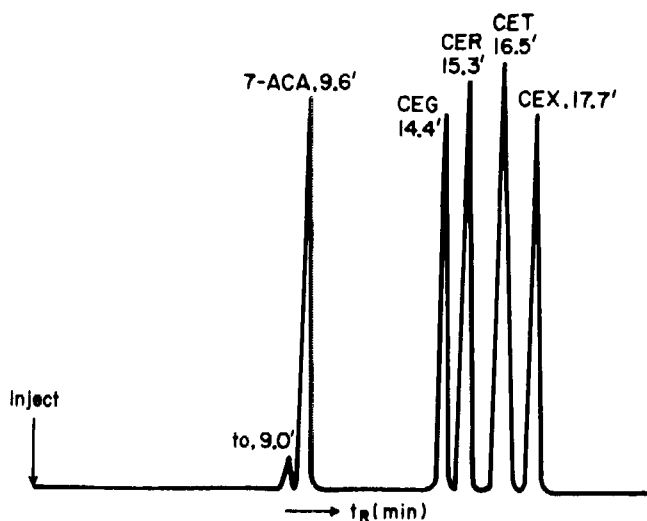


Figure 10. LC of Cephalosporins

Column, RP-2(4×30 cm); Mobile Phase, MeOH-H<sub>2</sub>O (1:2)

with PIC B-7; Flow Rate, 0.3 ml/min; Column Pressure, 1000 psi.

Correlation of Mobilities of Cephalosporin Anti-  
biotics in RPTLC and RPHPLC

Table 13 shows compares the separation behavior on RPTLC and RPHPLC described above. After correlation analysis between TLC  $hR_f$  values and HPLC R ( $R = t_0/t_R$ ), a correlation between the two separations was obtained, and the possibility of direct transfer of chromatographic systems from TLC to HPLC for separation of these antibiotics was confirmed.

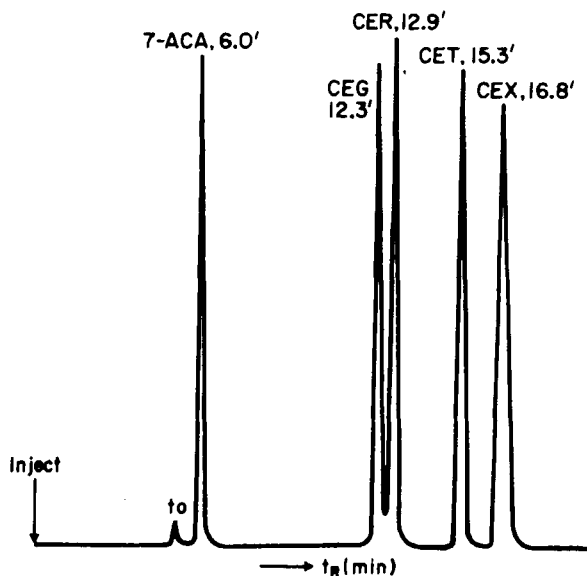


Figure 11. LC of Cephalosporins

Column, RP-2 (4x30cm); Mobile Phase, MeOH-H<sub>2</sub>O (1:4)  
with PIC B-7; Flow Rate, 0.5ml/min; Column Pressure, 1600 psi.

### CONCLUSION

#### Adsorption Chromatography of Steroidal Hormones

HPLC separation was investigated by adsorption chromatography of hardly separable stereoisomers such as 5:6-oxido-cholesteryl benzoates, 5:10-oxido-19-nor-androstane-3 $\alpha$ ,17 $\beta$ -diols and their 3,5-dinitrobenzoates, dexamethasone 21-acetate and betamethasone 21-acetate, and 5-cyano-19-nor-testosterones and their 3,5-dinitrobenzoates, and syn-,

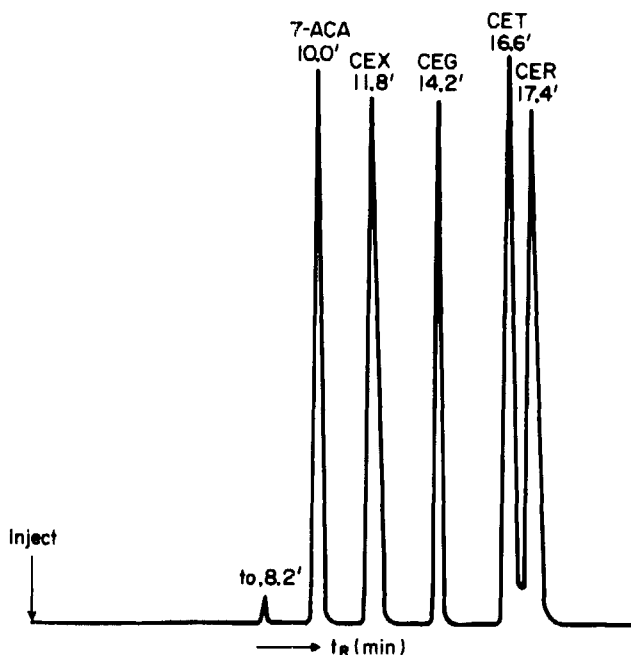


Figure 12. LC of Cephalosporins

Column, RP-2 (4×30cm); Mobile Phase, MeOH-H<sub>2</sub>O (1:2)

with PIC A; Flow Rate, 0.3ml/min; Column Pressure, 1200 psi.

anti-geometrical isomers such as  $\Delta^5$ -3 $\beta$ -hydroxyandro-  
stane-17-one oxime benzoates. Prior to HPLC with  
silica gel columns, silica gel TLC was performed to  
obtain suitable HPLC systems for separating of these  
compounds. Seven solvent systems for TLC did not  
give satisfactory separation in HPLC when used in  
the same ratios. As retention times ( $t_R$ ) and  
capacity factors ( $k$ ) of these isomers were small,  
the separation factors ( $\alpha$ ) were insufficient for

TABLE 13

Relationship of Mobility between Reversed-Phase  
TLC and Reversed-Phase HPLC with  
Cephalosporin Antibiotics

Compound	MeOH-H <sub>2</sub> O (1:4)	
	TLC, R <sub>f</sub> <sup>a)</sup>	HPLC, R <sup>b)</sup>
7-Aminocephalo- sporanic acid	0.93	0.90
7-Aminodesacetoxy- cephalosporanic acid	0.90	0.94
Cephalothin	0.70	0.90
Desacetylcephalothin	0.74	0.94
Cephalexin	0.61	0.63
Cephaloglycine	0.52	0.58
Cephaloridine	0.35	0.35

a) Merck silica gel 60 F<sub>254</sub> silanised (laboratory-prepared), Developing rate; 2 mm/min

b) Column: Merck RP-2 (10 μm, 4 mm x 30 cm, home-packed)

Flow rate: 0.3 ml/min

Linear Velocity: 2.38 mm/min

$$R = t_o/t_R$$

Regression analysis between TLC R<sub>f</sub> and HPLC R:

$$R = 1.0275R_f + 0.05, \quad \gamma = 0.915^{**}, \quad n = 7$$

good separation. For the sharp separation in HPLC, it was necessary to decrease the solvent strength of the same binary solvent mixture used for TLC had to be decreased, for instance from a 4:1 to a 1:1 ratio

in chloroform-n-hexane systems to separate 5:6-oxido-cholesteryl benzoates. The difference in solvent strength for efficient separation between TLC and HPLC might be attributed to the fact that in HPLC the solvent elution power acts in an isocratic manner while in TLC it acts in a gradient manner.

### Reversed-Phase Chromatography of Cephalosporin Antibiotics

HPLC separation was investigated by reversed-phase partition chromatography of eight kinds of cephalosporin compounds, 7-ACA, 7-ADCA, CEX, CET, CEG, CER, DACET and ECDACET. Prior to RPHPLC with dimethylsilanised silica gel columns packed by balanced density slurry method RPTLC with the same stationary phase was performed to obtain the suitable HPLC conditions for separating these compounds. Five solvent systems and five kinds of HPLC packings were tested: methanol-water, acetonitrile-water, acetone-water, dioxane-water and tetrahydrofuran-water; and aminopropyl-, cyanopropyl-, dimethyl-, octyl- and octadecyl-silanised silica gels. Among these, the combination of methanol-water and a di-

methylsilanised silica gel column gave satisfactory HPLC separation when the solvent ratio was 1:4.

The ion-pair chromatographic method was effective for sharp separation of CET from 7-ACA in HPLC. Correlation of mobility between TLC ( $hR_f$ ) and HPLC (R) was obtained [ $R = 1.05 R_f$  ( $n = 7$ ), correlation analysis:  $y = 1.0275x + 0.0513$ ,  $\gamma = 0.9153^{**}$ ,  $n = 7$ ], and the possibility of direct transfer of chromatographic systems from TLC to HPLC to separate cephalosporin antibiotics was confirmed.

#### REFERENCES

1. Duncan, G. R., J. Chromatogr., 8, 37, 1962.
2. Dahn, H. and Fuchs, H., Helv. Chim. Acta, 45, 261, 1962.
3. Loev, B. and Snyder, K. M., Chem. & Ind. (London), 1965, 15.
4. Komatsu, M. and Okano, S., YAKUGAKU ZASSHI, 87, 712, 1967.
5. Creis, W., Greenspan, A., Woodcock, T. and Gordon, C., J. Chromatogr. Sci., 14, 331, 1976.
6. Aoki, T., Konishi, H. and Hori, M., BUNSEKI KAGAKU, 23, 199, 1974.

7. Issaq, H. J., Shaikh, B., Pontzer, N. J. and Barr, E. W., *J. Liquid Chromatogr.*, 1, 133, 1978.
8. Hara, S. and Mibe, K., *Chem. Pharm. Bull.*, 23, 2850, 1975.
9. Golkiewicz, W., *Chromatographia*, 9, 113, 1976.
10. Majors, R. E., *Anal. Chem.*, 44, 1722, 1972.
11. Okumura, T., *BUNSEKI KAGAKU*, 26, 396, 1977.
12. Nagata, W., Yoshioka, M. and Okumura, T., *J. Chem. Soc. (C)*, 1970, 2365.
13. Nagata, W., Yoshioka, M. and Hirai, S., *Tetrahedron Letters*, 1962, 461.
14. Gondos, G., Matkovics, B. and Kovaes, O., *Microchem. J.*, 8, 415, 1964.
15. Okumura, T., *BUNSEKI KAGAKU*, 26, 848, 1977.
16. Eksborg, S. and Schill, G., *Anal. Chem.*, 45, 2092 (1973).